

Development of Equations for Estimating Greenhouse Gas Emissions from the Son La Hydropower Reservoir

Nguyen Thi The Nguyen¹, Pham Van Hoang², Nguyen Manh Khai^{3,*}

¹*Water Resources University, Tay Son, Dong Da, Hanoi, Vietnam*

²*Union of Science and Technology Vietnam, Xuan Dinh, Tay Ho, Hanoi, Vietnam*

³*VNU University of Science, 334 Nguyen Trai, Thanh Xuan, Hanoi, Vietnam*

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Abstract: Emissions of greenhouse gases such as CO₂ and CH₄ from artificial reservoirs, especially wide lakes in the tropics as the Son La hydropower reservoir, are leading to global warming. CO₂ and CH₄ gases in hydropower reservoirs are caused by the decomposition of organic matter in the lakes. In this study, regression analysis was used for estimating the relationships among water quality parameters measured at the Son La hydropower reservoir and the fluxes of greenhouse gas emissions from the reservoir. The regression analysis was also applied to develop regression equations predicting emissions of greenhouse gases from the lake. Results of study showed that the CO₂ emission from the Son La hydropower reservoir could be predictable from several water quality parameters of which 4 main factors are temperature, DO, alkalinity and pH. The amount of CH₄ emission from the Son La hydropower reservoir has solid relationships with 3 main factors, including temperature, COD and pH. The regression equations predicting CO₂ and CH₄ with the correlation coefficient of 0.93 and 0.92 have been tested with real data and gave the good results. Since, they could be introduced in reality.

Keywords: Greenhouse gas, hydropower reservoirs, water quality, regression equation.

1. Introduction

Energy sources which are generated from burning fossil fuel provide about 68% of global electricity in 2007 and are responsible for most of the anthropogenic greenhouse gas emissions to the atmosphere (accounts for approximately 40% [1]). Compared to fossil fuels, hydropower has been considered an attractive renewable energy source with the advantage of being less

harmful in terms of greenhouse gas emissions. Currently, hydroelectric power meets about 16% of the power supply of the world [2]. For countries which are dependent on hydroelectric energy, this kind of energy source accounts for 90%. Previously, hydroelectric energy are not considered as greenhouse gas emissions. However, recent studies showed that hydropower reservoirs could produce more carbon into the atmosphere than natural systems, especially in the first twenty years after flooding [3]. This is mainly due to the usually excessive availability of decomposable organic matter in hydroelectric reservoirs. Not only

*Corresponding author. Tel.: 84-913369778

Email: khainm@vnu.edu.vn

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large amounts of soil and terrestrial vegetation are flooded by damming rivers, but terrestrial organic matter derived from land erosion is continuously flushed into reservoirs as well. The usually high water residence time in reservoirs as compared to rivers, combined with high inorganic nutrient inputs, favors organic matter decomposition and, thus, the production of two major greenhouse gases – carbon dioxide (CO₂) and methane (CH₄). The amount of CO₂ and CH₄ emitted varies (a) among reservoirs (as function of drainage basin characteristics, reservoir morphology, climate, etc.); (b) within reservoirs (along longitudinal gradients from the tributaries to the dam, before and after the dam, etc.); and (c) over time (with reservoir aging, seasonally, daily, with changes in anthropogenic activities in the drainage basin, and with dam operation depending on energy needs and precipitation regime) [4]. Attempts to estimate the amounts of CO₂ and CH₄ emitted to the atmosphere should consider such variability which makes it a complex task. Today, there are at least 45,000 large hydroelectric reservoirs operating in the world [5]. The area of those lakes in the world is estimated at about 350.000km² [5]. The lakes which have large storage capacity need to be examined the impact on global warming.

The ever increasing global energy demand and the concern about the changes in environment have lead to an urge to assess the hydropower ‘footprint’ in terms of greenhouse gas emissions to the atmosphere. Since the early 90’s the role of hydroelectric reservoirs as sources or, as the opposite, sinks of greenhouse gases has rapidly become a global topic of investigation. The first studies of greenhouse gas fluxes from reservoirs focused on hydroelectric generation because it was, and still is, widely viewed as a carbon-free source of energy [6]. This view likely originated because before 1994, there were no data available on CO₂ and CH₄ emissions from reservoirs, even though it was well known that oxygen depletion resulting from active decomposition of flooded organic matter was common in waters of newly constructed

reservoirs. The first discussion of greenhouse gas emissions from reservoirs pointed out that greenhouse gas production per unit of power generated [6]. Then, there were many studies of greenhouse gas fluxes from reservoirs located in Canada [6], Brazil, Panama and French Guiana. Later, reservoirs in Finland, USA and Switzerland, China were studied. In the world until 2012, there were at least 85 research reports which focused on greenhouse gas from hydropower reservoirs [7].

In recent years, Vietnam has been facing growing manifestations of climate change. The natural conditions and especially the human activities including hydropower reservoirs have been caused impacts on the process of climate change. Following the Convention of the United Nations Framework on Climate Change (UNFCCC), Vietnam has established the National Communications (NCs) and Biennial Update Reports (BURs), including national inventory results on greenhouse gas emissions. Greenhouse gas emissions in Vietnam are estimated by following fields: energy, industrial processes, agriculture, land use changes and agricultural land use (LULUCF) and waste. So far, there is no official result for the inventory of greenhouse gas emissions in the field of hydropower.

The Son La hydroelectric reservoir, which is the largest one in Vietnam, has a catchment area of 43.760 km². It is also the largest reservoir in the field of capacity in Southeast Asia. To date, the Son La hydropower plant has been put into operation for about 5 years. Therefore it is necessary to access the possibility of greenhouse gas emissions from the reservoir and to set environmental management measures.

From the above requirements, this study was conducted to evaluate the possibility of greenhouse gas emissions and to develop equations for predicting the greenhouse gas emissions of CO₂ and CH₄ from the Son La hydropower reservoir. The research contributes to clarify the forecasting method of greenhouse gas emissions based on basic water quality

parameters in the Son La hydropower reservoir as well as other lakes located in the tropical areas. Currently, water quality monitoring is carried out periodically at hydropower reservoirs and it is done more favorably than that of CO₂ and CH₄. Thus the results of the study will help to take full advantage of periodically measured results of water quality following the Environmental Protection Law No. 55/2014 / QH13 2014 at the hydropower reservoirs to predict CO₂ and CH₄ emissions without continuous monitoring of those gases.

2. Study area and methods

2.1. Study area and object

The Son La hydropower plant is located at It Ong commune, Muong La district, Son La province. After seven years of construction, the Son La hydropower reservoir was inaugurated on December 23, 2012. The scale of the reservoir is as follows: the normal water level is 215m, the dead water is 175m, the installed capacity is 2,400 MW, the average power output is 9429 million kWh annually. The total reservoir capacity is 9260 million m³, the useful capacity is 6504 million m³. The catchment area of 43760 km² is located in three provinces of Son La, Dien Bien, Lai Chau. The lake has the largest width of about 1.5 km and 120km in length from the dam at the town of It Ong, Muong La district, Son La province to back up upstream at Lai Chau province. Diagram of the Son La hydropower reservoir is presented in Figure 1.

This paper focuses on CO₂ and CH₄ gases which are two major ones standing at the top of the list of greenhouse gases on the Earth. Besides, fundamental water quality parameters monitored periodically in the Son La hydropower reservoir related to greenhouse gas emissions are also taken into consideration.

2.2. Methods of study

2.2.1. Methods of sampling, sample preservation and determination of water quality

Sample collection, preservation and analysis of surface water quality carried out under the guidance of national technical regulations. The water quality parameters were analyzed including temperature, pH, TDS, conductivity, alkalinity, DO, COD, total nitrogen, PO₄³⁻. The water samples were collected at six locations as shown in Figure 1, in which the sampling locations C1, C2, C3, C5 are the effluents into the reservoir, C4 is in the middle of the reservoir and C6 is after the Son La dam. Sampling periods are the dry seasons (March) and the rainy seasons (August) in the years 2014 and 2015. The analysis was conducted at the laboratory of the Centre for Environmental Research, Institute Meteorology, Hydrology and Environment.

2.2.2. Sampling and determining methods of the greenhouse gases

Fluxes of greenhouse gases from water surfaces can be quantified using a number of techniques [8]. In this study, floating static chambers have been used to estimate the diffusive flux of CO₂ and CH₄ from the surface of reservoirs by calculating the linear rate of gas accumulation in the chambers over time.

CO₂ gas is collected following the method of air sampling in the sealed chamber Rolston (1986) [9], and is determined by applying the method under the ISO 5563-199. The size of CO₂ collecting box is as follows: the box diameter is 30 (cm), the box height is 20 (cm), of which the submerged part is 7cm, the useful height is 13cm. The air in the sealed container was sucked by the Kimoto -HS7 machine with the rate of 2 liters of gas per minute and is absorbed by Ba(OH)₂ solution. The air through the air receiver without CO₂ continues to return the sealed container to push the remaining CO₂ in the box. Sampling time is 10 minutes. CO₂ samples were collected at the same places and time with the water quality samples. After CO₂ is absorbed by Ba(OH)₂ solution, excess Ba(OH)₂ is titrated by oxalic acid.

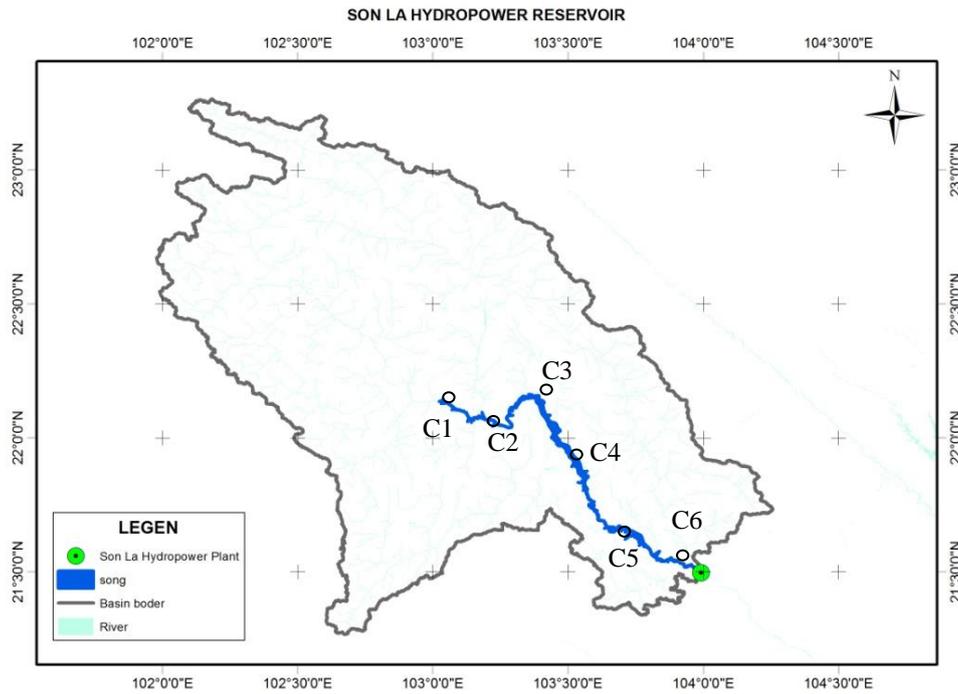


Figure 1. Location map of Son La hydropower reservoir and water quality collection points

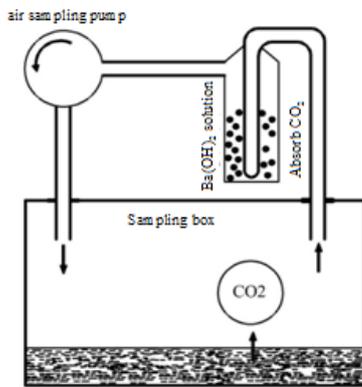


Figure 2. Sampling principle diagram of CO₂.

CH₄ gas is also collected following the method of air sampling in the sealed chamber Rolston (1986) [10]. The CH₄ collection box has the same size with the CO₂ collection box. The sealed chamber which has a determined area had been placed on the surface of the reservoir. The air was sucked by the air cylinder chamber at the time of 0 minute (in order to

determine the initial amount of CH₄ contained in sealed container), 10 minutes and 20 minutes. Gas samples were saved in neutral glass tubes with the volume of 20.0 ml. The air samples were analyzed by using gas chromatography machine GC17A and FID detector of which the carrier gas is N₂. CH₄ samples were collected and analyzed at the same places and times as the water samples.

2.2.3. Regression analysis technique

The regression analysis technique was used to develop the equations describing the relationships between water quality factors and CO₂, CH₄ gas emissions from the Son La hydropower reservoir. This study method has been being applied for forecasting in many fields like hydrological factors, climate, environment, economy ... The accuracy of the technique depends on the length of the data string. Multivariate regression equations have a general following form [9]:

$$Y_k = \beta_0 + \beta_1 X_1 + \beta_2 X_2 + \beta_3 X_3 + \beta_4 X_4 + \dots + \beta_k X_k; \text{ Correlation coefficient } R^2$$

Where:

- Y_k : dependent variable, k : number of independent variables

- X_i : independent variable

- β freedom coefficient, $\beta_{1,2,\dots,k}$: separate regression coefficients or slopes.

Correlation coefficient, R^2 , is always from 0 to 1. It is useful because it gives the proportion of the variance (fluctuation) of one variable that is predictable from the other variable. It is a measure that allows us to determine how certain one can be in making predictions from a certain model/graph. The correlation has low level when $0 \leq R^2 < 0.3$, average level when $0.3 \leq R^2 < 0.5$, quite close level when $0.5 \leq R^2 < 0.7$, high level when $0.7 \leq R^2 < 0.9$, very high level when $0.9 \leq R^2 \leq 1$.

In this study, dependent variables are CO_2 and CH_4 , while 9 independent variables are temperature, pH, TDS, conductivity, alkalinity, DO, COD, total nitrogen, PO_4^{3-} . Input data to develop the linear regressions of CO_2 and CH_4 are monitoring results of water quality at 6 locations in 4 periods in 2014 and 2015. In addition, periodically measurement data of water quality in the Son La reservoir in 5 years is also used for the study.

2.2.4. Data processing methods

The Excel and Eviews Software were used to statistically analyze the water quality results and to access links between greenhouse gas emissions in Son La and the water quality factors.

3. Results and discussions

3.1. Current status of water quality and greenhouse gas emissions from the Son La hydropower reservoir in the years 2014, 2015

The results of water quality analysis showed that most indicators of water quality in rainy

season had higher concentrations than those in dry season. The reason could be that during rainy season, higher water flows from the upstream of the basin carried more sediment, pollutants into the reservoir. Moreover, people living inside the basin took advantage of submerged land for crop cultivation, especially planting cash crops. When rainy season came, the agricultural waste and manure left over on this part submerged made the concentration of pollutants in the reservoir increasing. Compared to the National technical regulation on surface water quality (QCVN 08: 2008/BTNMT), water quality in the Son La reservoir was acceptable for purposes of irrigation, waterway or others.

The average CO_2 values emitting from the Son La hydropower reservoir in 2014 and 2015 fluctuated from 161.64 to 238.83 $\text{mg/m}^2/\text{day}$. The total CO_2 emission from the whole surface of the reservoir was about 36207.36 to 53497.92 tons/day, corresponding to 0.62 to 0.92 tons CO_2/MW . Compared to those values in some research in the world, for example the research on the Wohlen reservoir in Switzerland (the CO_2 value at the first year of operation was $1558 \pm 613 \text{ mg/m}^2/\text{day}$, dropped to $276 \pm 57 \text{ mg/m}^2/\text{day}$ at the 3rd year) and the Lungern reservoir in Switzerland (the CO_2 value was $136 \pm 353 \text{ mg/m}^2/\text{day}$ [11]), the level of CO_2 emission from the Son La hydropower reservoir after 5 years operation was moderate.

The average CH_4 value measured at the Son La hydropower reservoir in 2014 and 2015 ranged from 3.22 – 5.30 $\text{mg/m}^2/\text{day}$. The total CH_4 emission from the reservoir ranged from 153.44 to 1232 tons/day, corresponding to 0.0148 to 0.0213 tons CH_4/MW . Compared to some research findings on hydropower reservoirs (for example in China the CH_4 emissions in some lakes and reservoir were $2.88 \pm 1.44 \text{ mg/m}^2/\text{day}$, the value for the Three Gorge reservoir in China was about $7.2 \pm 2.4 \text{ mg/m}^2/\text{day}$ [12]), the level of CH_4 emission from the Son La hydropower reservoir was also moderate.

3.2. Evaluation of the relationships between greenhouse gas emissions with water quality parameters

3.2.1. The correlations between CO₂ and the water quality parameters

The correlation between CO₂ and the water quality parameters is shown in Figure 3 and

Table 1. The results show high correlations between CO₂ values and temperature (R² = 0.67), DO (R² = 0.55), alkalinity (R² = 0.65), pH (R² = 0.61). The correlation between CO₂ and conductivity is very low (R² = 0.06). This means that two variables have no relationship with each other. Therefore, the emission of CO₂ from the reservoir is affected primarily by temperature, DO, alkalinity and pH.

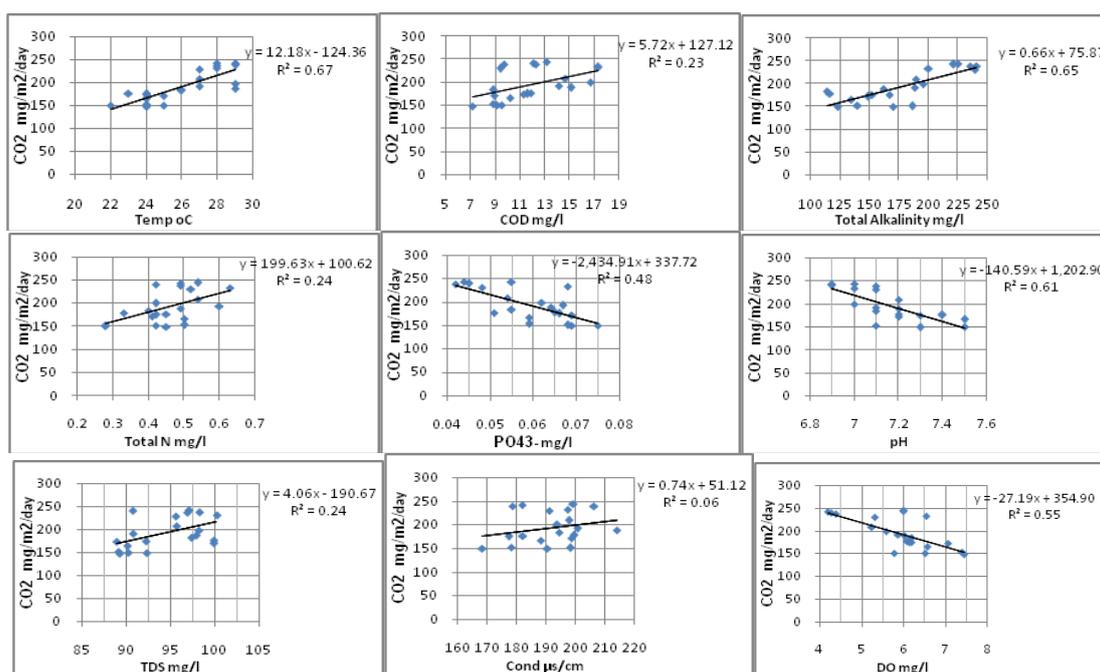


Figure 3. Correlations between CO₂ and temperature, pH, TDS, conductivity, alkalinity, DO, COD, total nitrogen and PO₄³⁻.

Table 1. Correlation between CO₂ and some water quality parameters

TT	Correlation	Function	R ²
1	CO ₂ and temperature	y = 12,18x - 124,36	0,67
2	CO ₂ and DO	y = -27,19x + 354,90	0,55
3	CO ₂ and COD	y = 5,72x + 127,12	0,23
4	CO ₂ and alkalinity	y = 0,66x + 75,87	0,65
5	CO ₂ and total nitrogen	y = 44,24x + 97,05	0,24
6	CO ₂ and PO ₄ ³⁻	y = 199,63x + 100,62	0,48
7	CO ₂ and pH	y = -140,59x + 1202,90	0,61
8	CO ₂ and TDS	y = 4,06x - 190,67	0,24
9	CO ₂ and conductivity	y = 0,74x + 51,12	0,06

CO₂ (mg/m²/day), temperature (°C), DO (mg/l), alkalinity (mg/l), total nitrogen (mg/l), PO₄³⁻ (mg/l), pH, TDS (mg/l) and conductivity (µs/cm.)

3.2.2. The correlations between CH₄ and the water quality parameters

The correlation coefficients R² of the dependent variable CH₄ and some water quality indicators are shown in Table 2 and Figure 4.

The results show high levels of correlation between CH₄ and temperature (R² = 0.6), COD (R² = 0.57), pH (R² = 0.58). Therefore, the emission of CO₂ in the reservoir is affected primarily by temperature, COD, pH.

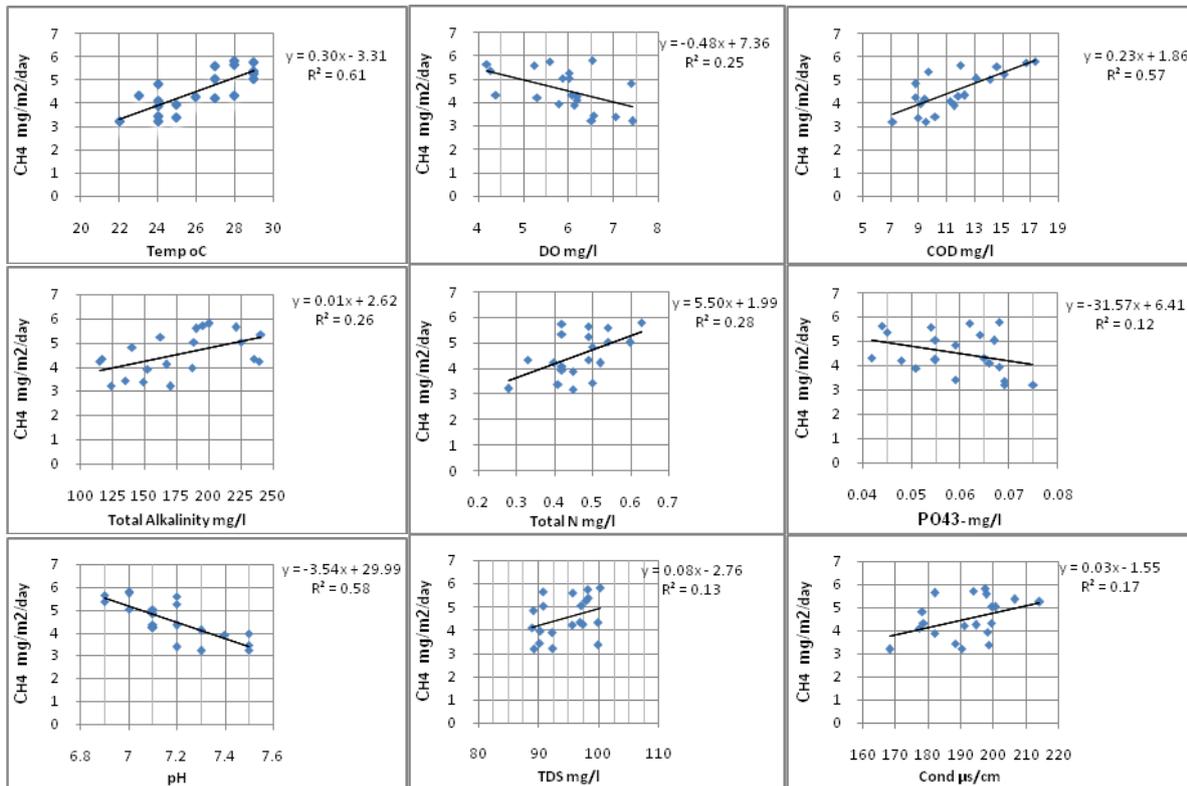


Figure 4. Correlation between CH₄ and temperature, pH, TDS, conductivity, alkalinity, DO, COD, total nitrogen and PO₄³⁻.

Table 2. Correlation between CH₄ and some water quality parameters

TT	Correlation	Funtion	R ²
1	CH ₄ and temperature	y = 0,30x - 3,31	0,61
2	CH ₄ and DO	y = -0,48x + 7,36	0,25
3	CH ₄ and COD	y = 0,01x + 2,62	0,57
4	CH ₄ and alkalinity	y = 0,01x + 2,62	0,26
5	CH ₄ and total nitrogen	y = 5,50x + 1,99	0,28
6	CH ₄ and PO ₄ ³⁻	y = -31,57x + 6,41	0,12
7	CH ₄ and pH	y = -3,54x + 29,99	0,58
8	CH ₄ and TDS	y = 0,08x - 2,76	0,13
9	CH ₄ and conductivity	y = 0,03x - 1,55	0,17

CO₂ (mg/m²/day), temperature (°C), DO (mg/l), alkalinity (mg/l), total nitrogen (mg/l), PO₄³⁻ (mg/l), pH, TDS (mg/l) and conductivity (µs/cm).

3.3. Development of predictive equations of CO₂ and CH₄ emissions from the Son La hydropower reservoir

3.3.1. The predictive equation of CO₂ emission

By applying the regression analysis technique and Eiview software, the forecasting equation of CO₂ emissions is as follows:

$$A1 = 367,62 - 3,04B - 9,508C + 1,33D + 0,28E + 85,17F - 662,45G - 46,07H + 2,55I \quad (1)$$

$$R^2 = 0,929$$

Where A1 = CO₂, B = temperature, C = DO, D = COD, E = alkalinity, F = total nitrogen, G = PO₄³⁻, H= pH, I = TDS.

The correlation between the dependent variable CO₂ and 8 independent variables (including temperature, DO, COD, alkalinity, total N, PO₄³⁻, pH and total dissolved solids) has the maximum correlation coefficient R² = 0,929. The value of correlation coefficient value depends on the independent variables. When the number of independent variables decreases, the R² also falls (see Table 3). This means that the predictive equation of CO₂ emission should be based on a certain number of water quality parameters to give the best results.

Table 3. The changes in the correlation coefficients between CO₂ with a number of water quality parameters

Number of parameters	Water quality parameters	R ²
8	Temperature, alkalinity, pH, DO, PO ₄ ³⁻ , total nitrogen, TDS, COD	0,929
7	Temperature, alkalinity, pH, DO, PO ₄ ³⁻ , total nitrogen, TDS	0,924
6	Temperature, alkalinity, pH, DO, PO ₄ ³⁻ , total nitrogen	0,867
5	Temperature, alkalinity, pH, DO, PO ₄ ³⁻	0,856
4	Temperature, alkalinity, pH, DO	0,847
3	Temperature, alkalinity, pH	0,808
2	Temperature, alkalinity	0,750
1	Temperature	0,670

Table 4. The changes in the correlation coefficients between CO₂ with a number of water quality parameters

Number of parameters	Water quality parameters	R ²
9	Temperature, pH, COD, total nitrogen, alkalinity, DO, conductivity, TDS, PO ₄ ³⁻	0,917
8	Temperature, pH, COD, total nitrogen, alkalinity, DO, conductivity, TDS	0,908
7	Temperature, pH, COD, total nitrogen, alkalinity, DO, conductivity	0,861
6	Temperature, pH, COD, total nitrogen, alkalinity, DO	0,857
5	Temperature, pH, COD, total nitrogen, alkalinity	0,840
4	Temperature, pH, COD, total nitrogen	0,839
3	Temperature, pH, COD	0,838
2	Temperature, pH	0,680
1	Temperature	0,612

3.3.2. The predictive equation of CH₄ emission

By applying the same process with CO₂, the forecasting equation of CH₄ emission has the following form:

$$A2 = 29,44 - 0,03B + 0,11C + 0,20D + 0,00087E - 1,24F - 21,76G - 3,07H - 0,09I + 0,028K \quad (2)$$

$$R^2 = 0,917$$

Where A2 = CH₄, B = temperature, C = DO, D = COD, E = alkalinity, F = total nitrogen, G = PO₄³⁻, H= pH, I = TDS, K = conductivity.

The maximum correlation coefficient between the dependent variable CH₄ and the 9 independent variable (including temperature, DO, COD, alkalinity, total N, PO₄³⁻, pH, total dissolved solids and conductivity) is 0.917. The reduction of number of water quality parameters makes the R² decreasing (Table 4). Like CO₂, the predictive equation of CH₄

emission should be based on a certain number of water quality parameters to give the best results.

3.4. Verification of the predictive equations of CO₂ and CH₄ emissions from the Son La hydropower reservoir

In order to verify the predictive equations of CO₂ and CH₄ emissions, the equations (1) and (2) above are applied to calculate the amount of CO₂ and CH₄. The input data is the measured values of water quality in 4 stages in the years 2014, 2015 at 6 locations (Figure 1). The results of statistical analysis are presented in table 5 and figure 5. As can be seen in those table and figure, the predictive values of CO₂ and CH₄ emissions by the equations are slightly higher than the experimental values. The results show the same tendency as observed in nature. Therefore, they can be applied to estimate the greenhouse gas emissions from the Son La hydropower reservoir.

Table 5. Statistical analysis of fluxes of CO₂ and CH₄ at the Son La hydropower reservoir

Parameters	Values of CO ₂ (mg/m ² /day)		Values of CH ₄ (mg/m ² /day)	
	Measured	Calculated	Measured	Calculated
Number of sample (n)	24	24	24	24
Minimum value	149,92	136,61	3,21	3,34
Maximum value	245,72	243,60	5,82	5,93
Average value	193,45	194,14	4,54	4,74
Standard deviation	33,85	32,70	0,87	0,83

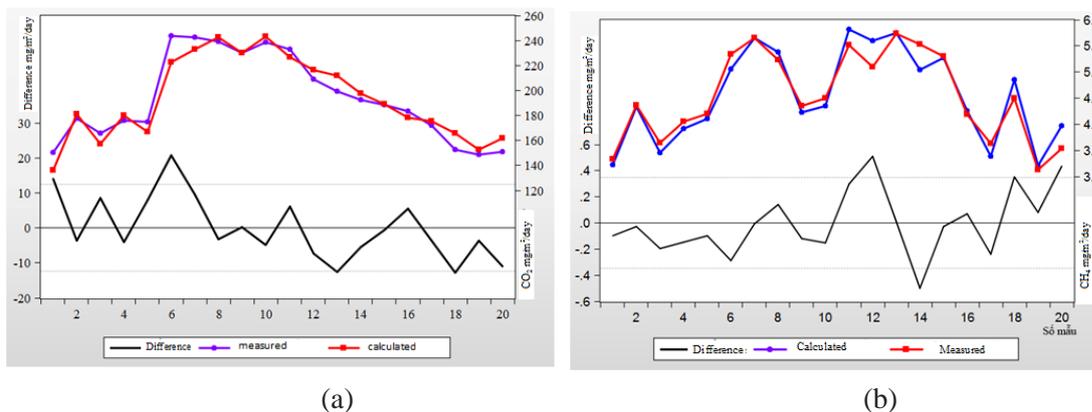


Figure 5. The calculated and measured fluxes of CO₂ (a) and CH₄ (b) at the Son La hydropower reservoir.

4. Conclusion

The amounts of CO₂ and CH₄ greenhouse gas emissions from the Son La hydropower reservoir were average compared to other reservoirs in the world. CO₂ emission from the Son La hydropower reservoir has relationships with several water quality parameters including 4 main factors: temperature, DO, alkalinity and pH. The amount of CH₄ emission from the reservoir also has relationships with several water quality parameters including 3 main factors: temperature, COD, pH. The regression equations predicting emissions of CO₂ and CH₄ in the Son La hydropower reservoir have been developed upon the actually measured values of water quality at the reservoir and give fairly consistent results with reality. Therefore, those equations can be used to estimate the amounts of CO₂ and CH₄ based on the periodic measurement of water quality. They also give a basis for making management measures to reduce greenhouse gas emissions from the reservoir in a better way.

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Nghiên cứu xây dựng phương trình ước tính lượng khí nhà kính cho hồ thủy điện Sơn La

Nguyễn Thị Thế Nguyên¹, Phạm Văn Hoàng², Nguyễn Mạnh Khải³

¹Trường Đại học Thủy lợi, Tây Sơn, Đống Đa, Hà Nội, Việt Nam

²Liên hiệp Khoa học và Công nghệ Môi trường, Xuân Đình, Tây Hồ, Hà Nội, Việt Nam

³Trường Đại học Khoa học Tự nhiên, ĐHQGHN, 334 Nguyễn Trãi, Thanh Xuân, Hà Nội, Việt Nam

Tóm tắt: Phát thải khí nhà kính như CO₂ và CH₄ từ các hồ chứa nhân tạo, đặc biệt là các hồ lớn ở vùng nhiệt đới như hồ thủy điện Sơn La đang dẫn đến sự nóng lên toàn cầu. Khí CO₂ và CH₄ trong các hồ thủy điện sinh ra do sự phân hủy các chất hữu cơ trong lòng hồ. Trong nghiên cứu này, các thông số chất lượng nước như nhiệt độ, DO, COD, TDS, pH, tổng nitơ, phosphat, tổng độ kiềm, độ dẫn điện đo được tại hồ thủy điện Sơn La đã được phân tích hồi quy để tìm mối tương quan của chúng với lượng khí nhà kính phát thải từ hồ chứa này và từ đó xây dựng phương trình hồi quy dự đoán lượng khí nhà kính phát thải từ hồ. Kết quả phân tích hồi quy cho thấy lượng khí CO₂ phát thải từ hồ thủy điện Sơn La có mối quan hệ với nhiều thông số chất lượng nước trong đó có 4 yếu tố chính là nhiệt độ, DO, độ kiềm, pH. Lượng khí CH₄ phát thải từ hồ thủy điện Sơn La có mối quan hệ 3 yếu tố chính là nhiệt độ, COD, pH. Phương trình hồi quy dự đoán lượng khí CO₂ và CH₄ với hệ số tương quan là 0.93 và 0.92 đã được kiểm định với số liệu thực tế và cho kết quả khá tốt, từ đó có thể đưa vào áp dụng trong thực tế.

Từ khóa: Khí nhà kính, hồ thủy điện, chất lượng nước, phương trình hồi quy.