# PREPARATION OF Pb(Zr,Ti)O<sub>3</sub> THIN FILMS SPUTTERED FROM A MULTI-ELEMENT METALLIC TARGET

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Abs:ract: PZT thin films have been grown on Pt/Ti/SiO<sub>2</sub>/Si substrates by the reactive RF-magnetron sputtering deposition method using a multi-element metallic target. The ABO<sub>3</sub> perovskite structure in the thin films sputtered at a substrate temperature of 250 °C was formed through various intermediate phases by annealing at 650 °C for 1 hour in the air or an oxygen gas ambience. The film compositions were estimated by Electron Probe Micro-Analyzer (EPMA). The surface morphology observed by Atomic Force Microscopy (AFM) showed a densely packed grain structure with no rosettes structure. The remanent polarization value of the thin film with a thickness of 500 nm was  $1.37 \,\mu\text{C/cm}^2$ .

# 1. Introduction

Over the last 20 years, thin films of lead zirconate titanate,  $Pb(Zr,Ti)O_3$  (PZT), have been attracted much attention due to their applications in nonvolatile memory and MicroElectroMechanic Systems (MEMS) devices[1,2]. Most of works have been concentrated on the compositions of the Morphotropic Phase Boundary (MPB) at around Zr/Ti=50/50 compositional ratio, at which the piezoelectric and ferroelectric properties show their maximum values. Various technical solutions proposed for obtaining the thin films of PZT such as sol-gel method, pulsed laser ablation, sputter deposition, and ion beam sputtering have been employed. Among them, the sputtering method has been most widely used for obtaining high quality PZT films because of its relatively simple fabrication process.

One of the most important factors in the preparation of PZT thin films is the control of the correct perovskite phase formation in order to obtain the desired electrical properties. Mostly, in the as-deposited films there have been no perovskite phase. So post-deposition annealing treatments at high temperatures, typically in the range 600-700  $^{\circ}$ C for 1 hour or more, are necessary to crystallize the film into perovskite structure.

In this study, PZT thin films were fabricated by the reactive RF-magnetron sputtering method. The perovskite phase crystallization of PZT films was investigated in term of the effects of annealing time and heat treatment ambience by X-Ray Diffraction (XRD). The surface morphology and electrical properties of the films were also reported.

# 2. Experimental procedure

SCM-400 13.56MHz RF-An Alcatel magnetron sputtering system was used to deposit the PZT thin films on Pt/Ti/SiO2/Si substrates. During sputtering multiplayer process, the substrates, as bottom electrodes, were heated at 250 °C. The sputtering target was multi-element metallic, the design of which is shown in Fig. 1. This was composed of individual sectors of Pb, Zr, and Ti metals with Pb/Zr/Ti:2/7.5/8.5 compositional ratio. Prior to the deposition, the sputtering chamber was pumped down to a base pressure of  $10^{-6}$  mbar.



Fig. 1. The multi-element PZT metal target (after sputtering)

Subsequently, gas mixture of Ar and  $O_2$  with Ar/O<sub>2</sub>:60/40 ratio was introduced to a pressure of  $2x10^{-2}$  mbar. An RF-power was fixed at 200 W. The selected sputtering conditions for the deposition of bottom electrodes and PZT thin films are summarized in Table 1.

	Bottom electrodes	PZT thin films
RF- power	200 W	
Base pressure	~10 <sup>.6</sup> mbar	
Substrate	SiO <sub>2</sub> /Si	Pt/Ti/SiO <sub>2</sub> /Si
Target	Ti	Pb/Zr/Ti
	Pt	
Working pressure	3x10 <sup>-3</sup> mbar	1x10 <sup>-2</sup> mbar
Ambience gas	Ar	O <sub>2</sub> /Ar:40/60
Substrate temperature	T <sub>Room</sub>	250 °C
Deposition time	Ti: 5 min.	4 hours
	Pt: 30 min.	
Film thickness	Ti: 20 nm.	500 nm
	Pt: 100 nm	

Table 1. The conditions for preparation of bottom electrodes and the PZT(51/49)thin films.

The PZT film thickness was determined about 500 nm by the grazing incident X-Ray Reflection (XRR), as illustrated in Fig. 2. The Zr/Ti:51/49 compositional ratio of the PZT films was analyzed using Electron Probe Micro-Analyzer (EPMA). In order to crystallize perovskite phase, the as-deposited films were annealed using

conventional furnace at 650°C fixed temperature in the air and an oxygen ambience for 30, 60, 90, 120 min., Phase and crystalline respectively. behavior analyses of PZT films were performed using X-Ray Diffractometer with the CuK<sub> $\alpha$ </sub> source ( $\lambda$ =1.5405 Å). Surface morphology and grain size were Atomic Force investigated using The ferroelectric (AFM). Microscopy property of PZT thin films were also measured using RT66A standardized hysteresis tester (Radiant Technology).

#### 3. Results and discussion

#### Crystallographic structure

The crystal structure of the as-deposited and annealed PZT thin films was examined by XRD with patterns in comparison crystallographic information reported earlier on PZT films and ceramics. This part may shed light on the optimum annealing time and ambience required for the perovskite PZT formation. The XRD patterns of the films annealed for the various times in the air are given in Fig. 3.

It can be seen from the figure that all of films always show two peaks at  $2\theta=40^{\circ}$  and  $46.55^{\circ}$ corresponding to the Pt polycrystalline phase.

In case of the as-deposited



thin film annealed at 650°C for 60 min



Fig. 3. The XRD pattern of PZT thin films annealed for various annealing times

film, there are two peaks at 29.6° and 34.15° depicting non-ferroelectric pyrochlore,  $Pb_2Ti_2O_6$  cubic oxygen deficiency phase at low temperature. A peak at 56.24° is identified  $\alpha$ -PbO<sub>2</sub> structure. Mean while it is possible that  $ZrO_2$ ,  $TiO_2$  are presented in an amorphous form because whole Pb, Zr, Ti were oxidized in sputtering process simultaneously.

When the film was annealed for 30 min.,  $\alpha$ -PbO<sub>2</sub> transferred to the lead-rich phases PbO<sub>ss</sub> that are observed at  $2\theta$  of  $23.2^{\circ}$ ,  $25.45^{\circ}$ ,  $32.1^{\circ}$ , and  $42.4^{\circ}$ . Their relative high intensity may show the excess contents of Pb and O<sub>2</sub> in the thin film[3]. The peaks at  $29.05^{\circ}$  and  $33.94^{\circ}$  match with the pyrochlore monoclinic AB<sub>3</sub>O<sub>7</sub> phase, which is normally caused by a deficiency in lead due to an evaporation at high temperature. In this work, the pyrochlore phase appears even when lead-rich PbO<sub>ss</sub> compound presented in accordance with [4]. It has revealed that the formation of AB<sub>3</sub>O<sub>7</sub>-type was favored over ABO<sub>3</sub>-type even when the Pb/Ti ratio was greater than one and AB<sub>3</sub>O<sub>7</sub> phase seemed to be an intermediate phase before the PZT films crystallized into the ABO<sub>3</sub> perovskite phase. The peaks at  $21.75^{\circ}$  and  $38.55^{\circ}$  perform the perovskite struture. It is confirmed that the PZT films deposited at the substrate temperature  $250^{\circ}$ C can be converted into the perovskite crystalline structure at the annealing temperature  $650^{\circ}$ C with a sufficient annealing time.

Increasing the annealing time to 60 min., complete perovskite crystallization occurs. Besides the former peaks, the others at  $30.77^{\circ}$ ,  $44.15^{\circ}$ , and  $54.85^{\circ}$  according to the perovskite phase appears. The ratio of intensity between the peaks show that the structure of the PZT thin film is polycrystalline (like PZT ceramic)[5]. However, the (110) intensity peak at  $30.77^{\circ}$  is largely superior to the others, thus it is considered that the thin films has a preferred-(110) orientation.

The trend suggests that these films may be converted into the single perovskite phase by increasing the annealing time or annealing temperature. Nevertheless, the annealed films for more than 60 min. are diphase with pyrochlore and perovskite coexistence. The pyrochlore peaks reappear that proves Pb loss. The deficiency of Pb is understandable by considering its high volatility for a long annealing time at a high temperature. After the annealing time reaching 120 min., the non-ferroelectric pyrochlore phase is predominating with strong and sharp peaks.

To investigate the effect of annealing ambiences, the PZT thin films were annealed for 60 min. at 650 °C in the air, an oxygen and an argon gas ambience and the results are depicted in Fig. 4. As above discussion, the thin film, which was heated in the air, has the polycrystalline perovskite structure with predominated (110) orientation. The film annealed in the oxygen ambience has still the correct perovskite structure but (100) orientation is preferred. It may be caused by the influence of oxygen content on PZT orientations. It has been reported that the oxygen partial pressure is an important factor in determining the Pb valence state in the PZT films, which dictates the kinetics of the pyrochlore-perovskite phase transformation[6]. A low Pb valency state enhances the kinetics of transformation, whereas a high Pb valence state suppresses the kinetics of transformation leading to incomplete transformation of the pyrochlore to perovskite phase. The growth of PZT(100) on Pt/Ti/SiO<sub>2</sub>/Si wafers attributed to the formation of crystalline PbO(001) during prolysis. PbO has good lattice mitching with PZT(100) orientatica; this lowers the interfacia energy and promotes the nucleation of PZT(100). On the contrary, for the PZT film annealed in the argon ambience the pyrochlore phase dominates completely.

Although PZT films were annealed for different times in various ambiences, the (110) or (100) orientation is mainly observed. These results imply that PZT essentially has natural preference to grow in (110) or



Fig. 4. The XRD pattern of PZT thin films annealed in various annealing ambiences.

(100) directions at temperatures less than 700  $^{\circ}C[7]$ .

On the basic of this information, the crystallization of sputterd PZT thin films was stucied. The as-deposited film shows peaks due to  $\alpha$ -PbO<sub>2</sub> and A<sub>2</sub>B<sub>2</sub>O<sub>7·x</sub> crystalline while ZrO<sub>2</sub> and TiO<sub>2</sub> are amorphous. After annealing, lead-rich PbO<sub>ss</sub> and lead-deficient AB<sub>3</sub>O<sub>7</sub> phases were found as the intermediate phases and they reacted each other to form the perovskite structure during post-deposition heat treatments. In order to obtain the correct perovskite crystal structure the PZT thin films need to be annealed for 60 min. at 650 °C in the air or the oxygen ambience.

## Surface morphology and grain size

surface morphological The investigation on the sputtered PZT thin film annealed at 650 °C for 60 min. in the air was shown in Fig. 5. The film has a smooth surface with no relative microcrack, a dense structure, and fine grain distributing homogeneous. Agreeing well with AFM result, the average grain size is also estimated to be 50 nm by using Sherrer's equation with the halfwidth of the (110) diffractive peak observed at  $2\theta$ =30.77° in the XRD pattern in Fig. 4.



Fig. 5. The AFM micrograph of PZT thin film annealed at 650°C for 60 min. in the air

## Ferroelectric property

Ferroelectric behavior of the PZT film annealed at 650 °C for 60 min. in the air was studied by a plotting P-E hysteresis loop and is shown in Fig. 6.

The remenant polarization  $P_r$  of the PZT film is about  $1.37 \ \mu C/cm^2$ . The film exhibits the characteristic "elliptic" shaped ferroelectric response of a hard ferroelectric. This shape is common for film referred to as "lossy" or "leaky", which is not good insulator. The loop has shifted toward the negative side. This asymmetry also observed by many workers was attributed to the difference interfacial states between in top electrode/PZT interface and PZT/bottom electrode interface[8] due to the electrodes were prepared in completely different conditions.



Fig. 6. The P-E pattern of the PZT thin film at 650 °C for 60 min. in the air

# 4. Conclusions

PZT thin films were prepared on platinized silicon substrates by RFmagnetron sputtering method using a multi-element metallic target. Lead-rich  $PbO_{ss}$  and lead-deficient  $AB_3O_7$  phases were discussed as immediate phases in the  $ABO_3$  correct perovskite crystallization. The complete perovskite structure could be obtained at the optimized condition being at 650 °C annealed temperature for 60 min. in the air or the oxygen ambience.

# References

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